

Chemical Composition and Antioxidants of Lepidium Sativum and L. aucheri

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Abstract— Antioxidant activity of total glucosides contents of the extract of Lepidium sativum and Lepidium aucheri leaves was determined using free radical scavenging activity 2,2-Diphenyl-1-Picrylhydrazyle (DPPH) by adding different concentrations of glycoside to DPPH. The inhibitory activity determined by using five different concentrations of glycolysis of L. sativum and L. aucheri leaves extracts. The results indicated that 1000 mg/ ml concentration showed radical scavenging activity as strong as than low concentrations. Out of the two species L. aucheri had the greatest abundance of antioxidant compared with L. sativum. The inhibition percentage of L. sativum as found to be 78.211 in 1000 µg/mL, was comparatively lower than of standard ascorbic acid and L. aucheri extract 91.972 and 97.018 µg/mL, respectively. L. aucheri extract has strong activity that reach almost as high as ascorbic acid.

Total antioxidant capacity of the test samples was calculated using the standard line as ascorbic acid equivalents (AAE) per gram of the leaves extract, of ascorbic acid (y = 0.0629x + 48.356, $R^2 = 0.4085$). The results of *L. sativum* and *L. aucheri* was (y = 0.0436x + 43.48, $R^2 = 0.3493$) and (0.0648x + 48.761, $R^2 = 0.3483$) respectively. The two species with various concentration showed the strongest antioxidant activity with its significantly smaller IC₅₀ values,the best exhibited a quite recorded in *L. aucheri* (IC₅₀ =19.12 µg/ mL, followed by *L. astivum* 149.541 µg/mL. compared with ascorbic acid 26.136 µg/ mL.

The GC analysis indicated that L. sativum had high number of glycoside compounds 36 components compared with L. aucheri 19 components. The major components of L. sativum were found to be: Benzyl nitrile (22.24 %), N,N-Dimethylaminoethanol (17.53%), 2-Hydroxy-1-(1'-pyrrolidiyl)-1-buten-3-one (11.08 %), D-Proline (7.33 %), Butyrolactone (4.97%) and 1-(1'-pyrrolidinyl)-2-propanone (4.14), GC -MS analysis of L. aucheri represented DL-Proline, 5-oxo-, methyl ester (42.26%); 2-Furancarboxaldehyde, 5-methyl- (12.64%); 2,3-dihydro-3,5-dihydroxy-6-methyl-4H-Pyran-4-one, (9.87%); Hexadecanoic acid, 2-hvdroxv-1-(hydroxymethyl)ethyl ester (2.33%); 2-Methoxy-4-vinylphenol (2.02%).

Keywords— Antioxidant activity, GC-MS, *L. sativum*, *L. aucheri*, Chemical compounds.

I. INTRODUCTION

The Cruciferae (Brassicaceae) is one of the largest families of Angiosperms, also known as

the mustard family. The family mustard is widespread in the world, especially in the Mediterranean region, it was grown in the cold and temperate regions of the northern hemisphere, which includes about (338 -500) genera and (3000-2000) (Al-Mayah, 2001; Al-Shehbaz; 1984, Al-Musawi, 1987; Ait-yahia *et al.*, 2018). In Iraq, they have almost 80 genera and more than 177 species spread over 10 Tribes (Townsend, and Guest, 1980).

L. sativum L. and L. aucheri are a small, annual herb, known as 'Rhashad' in Iraq but in some regions called Garden cress, pepperwort, pepper grass, While L. aucheri called 'Rhashad barri'. L. sativum are widely used in Arabic countries for their medicinal properties. L. sativum about 15- 50 cm in height, while L. aucheri up to 30 cm high, L. sativum is glabrous, contains branches on the upper parts, the flowers are clustered in branched racemes, white to pinkish flowers. L. aucheri recognized the branches and whole plant with purple tinge and few leaves, fruit small siliculas (Facciola, 1990; Abdel karim et al., 2017; Hussein et al., 2017; Singh and Singh, 2018).

L. sativum and L. aucheri are have several and traditionally properties. Literature search reported that analgesic, aperient, anti-anemic activity, alternative, exhibit oral an contraceptive, antihypertensive effect, Tachyphylactic, antispasmodic, hypoglycemia antidiarrheal, laxative, antirheumatic, anticoagulant, antiinflammatory, anticancer activity; asthma, cough, diarrhea, dysentery, skin disease, aphrodisiac and carminative properties (Ahsan et.al.1989; Patel et al., 2009; Maghrani et al., 2005; Mullen et al., 2007; Ravi et.al., 2011; Kbyeh , 2015; Abdel

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karim et al., 2017). Hussein et al (2017) reported that L. sativum is useful in swelling, stiffness, anti-diarrheal, cardiotonic, hypotensive, antimicrobial, bronchodilator, and hypoglycemic applications. L. sativum young leaves are eaten raw or cooked, while its seeds are used in peppery flavor. The seeds are used in treatment of chronic liver enlargement and spleen diseases. (Usher, 1974; Merzouki et al., 2000; Mullen et al., 2007; Datta et al., 2011; Fan et al., 2014; Xiao et al., 2014; Hussein et al., 2017; Singh and Singh, 2018).

phytochemical studies of L. sativum showed presence of alkaloids, flavonoids, fatty acids, coumarins, flavonols, glycosylation (mono, di and triglycoside). sulphur glycosides, glucosinolate) triterpenes, sterols, phosphorus, thiamine, and niacin, sinapoyl malate, sinapic acid, sinapine and various imidazole alkaloids (Abdel karim et al.,2017; Hussein et al., 2017; Singh and Singh 2018). As well as contains amount of uric acid, iron, carotene, calcium, folic acid beside vitamins A and C. As well as, it contains protein; leucine; glutamic acid and methionin. glucosinolates are the major secondary compounds. Malar et al., (2018) reported that the seeds of L. sativum contains two glucosinolates, glucotropaeolin and (Singh gluconasturin and Singh, 2018). Flavonoids (flavones, flavanones, flavonols. isoflavones. flavanols, chalcones and anthocyanins) are present in this plant and they are commonly found conjugated to sugars in the form of O-glycosides or C-glycosides forms isothiocyanates, formed which is with glucosinolates (Kassie et al., 2002; Al-Snafi, 2019).

In recent years GC-MS analysis have been proved a valuable method for the identified of components (Hussein et al., 2017). The GC-MS spectrum revealed 19 compounds in methanolic extract of L. sativum leaves such as Glycerin, Furfural, Allyl isothiocyanate. While methanolic seeds extract showed the presence of 46 compounds included 4H-pyran-4-one, 2furancarboxaldehyde thiocyanic acid, 2-methoxy-4vinylphenol, d-mannose (Al-Snafi, 2018). 15 compounds identified by GC-MS analysis of total alkaloidal extract from L. sativum seeds reported by (Singh and Singh, 2018). 83 compounds were identified of methanol extract of seeds (Benzyl nitrile, Squalene, Hexanedioic acid, dimethyl ester and Azulene (Abu-Rumman, 2018). Lee and Chang (2019) recorded 25 chemical compounds in methanol extract of *L. meyenii*.

Medicinal plants are important sources of antioxidant. Brassicaceae crops are among those plants that have the highest antioxidant activity (Wang and Zheng, 2001; Soengas *et al.* 2012). reported that phenolic content gives a strongest and the highest antioxidant activity. Ethanolic extract of *L. sativum* leaf, shoot and stem were studied against DPPH, high scavenging activity was observed in the shoot (12.19±02%). (Malar *et al.*, 2018). Flavonols have antioxidant activity and important in inhibition of cancer cell proliferation (Ait-Yahia *et al.*, 2015). Significant antioxidant activity noticed of methanol extract of *L. sativum* subsp *spinescens* and *Lepidium meyenii* Walp (Selek *et al.*, 2018; Lee and Chang, 2019).

Antioxidant activities by (DPPH) freeradical-scavenging assays for EC50 were determined as 330.99 (Dadas) and 346.65 (Izmir) (Sat et al., 2013). The IC50 for scavenging DPPH was 0.61 mg/ml of Lepidium meyenii (Sandovala et al., 2002). While seeds extract has good DPPH radical scavenging activity and IC (Eddouks et al., 2005). The Oz (2011) determined that antioxidant capacity of a L.sativum as EC50 was 233.475 mg/ml. The maximum increase in antioxidant activity in petroleum ether extract was observed 3.125 mg/ml to 6.25 mg/ml (Bhasin et al., 2011). The antioxidant activity of the methanol extract of L. sativum found that the IC50 values are 62µg/ml (Ahamad et al., 2015) and IC50 of 925.22±0.02 ppm (Chatoui et al., 2016). Ethanol extract showed concentration-dependent antioxidant activity (0.146 to 18.75 mg/ml) (Al-Snafi, 2019).

The aim of this study was a quantity - quality analysis of the glycoside composition by using GC-MS analysis and antioxidant properties of glycoside extracts of *L. sativum* and *L. aucheri* leaves.

II. . MATERIALS AND METHODS

1- Plant collection

Lepidium sativum L. and *L. aucheri* leaves were collected from the Basrah city - Iraq in March 2019. The leaves were identified according to the flora of Iraq. Leaves air dried until used.

2- Determination of total glucosides contents

Glycosides extraction: Method of (Harborne, 1984) with some modification was

followed: 25 gm of plant powder added to 250 ml of ethyl alcohol (70%), leave mixture for 24 hours in the magnetic stirrer and filtrated for getting ethanol extract. extract was concentrated by rotary evaporator, 50ml of n-butanol were added to the mixture by separation funnel with shaking and pulled the water layer. The process was repeated three times and dried water drawn layer at a temperature of $30C^{\circ}$ until full drying and then put the extracts in tubes with lid and kept freeze until use.

3- Determination of antioxidant activity by DPPH assay:

The antioxidant activity of L. sativum and L. aucheri leaves glycosides was determined by DPPH assay using (Hatano et al., 1988) with some modifications. The L. sativum and L. aucheri glycoside extract was prepared in various concentrations (0, 10, 25, 100, 500) µg/ml diluted with methanol. 0.004 mg from DPPH was dissolved in 100 ml of methanol. The absorbance at 517 nm was determined against control after an incubation for 30 min at room temperature by using a spectrophotometer. DPPH (50 µg/ml) was used as the control, ascorbic acid as the standard in triplicate for the standard. The antiradical activity was revealed as IC₅₀ of DPPH scavenging activity by observing the 50% inhibitory concentration for extract using the calibration curve.

Percentage of antioxidant activity of free radical DPPH was calculated as follows:

Antioxidant activity (Inhibition) % = $[(A_{control} - A_{sample}) / A_{control}] \times 100$ Where: $A_{control}$ is the absorbance of the control reaction and A_{sample} is the absorbance in the presence of extract.

4- GC–MS analysis of glycosides

GC-MS analysis was carried out by using a Shimadzu GC-QP 2010 ultra-gas chromatograph. The GC oven temperature was programmed from 40°C to 280°C at hold of 10 C/min. Helium was used as a carrier gas. The pressure was 7.0699 psi. The column flow was 1mL/min, and purge flow 3 ml/min the injector temperature was 290°C with split injection mode. The MS scan conditions incorporate the following: source temperature, 200°C; interface temperature (MSD transfer line), 290°C; solvent cut time 4 min, scan speed, 1562 (N2); range 35 m/z to, 650 m/z. Chemical compound of the *L. sativum* were identified by comparing the spectra with known compounds stored in the NIST library (2005).

III. RESULTS AND DISCUSSION

1- Antioxidant assays of *L. sativum* and *L. aucheri* using DPPH radical- scavenging:

Antioxidant activity of total glucosides contents of the extract was determined using free radical scavenging activity (DPPH) by adding different concentrations of glycoside to DPPH. The inhibitory activity determined by using five different concentrations of glycolysis of *L. sativum* and *L. aucheri* leaves extract are represented in (Figure 1).

The results indicated that 1000 mg/ ml concentration showed radical scavenging activity as strong as than low concentrations (Figure 1). Out of the two species *L. aucheri* had the greatest abundance of antioxidant compared with *L. sativum*. The inhibition percentage of *L. sativum* as found to be 78.211% in 1000 μ g/mL, was comparatively lower than of standard ascorbic acid and *L. aucheri* extract 91.972% and 97.018% respectively (Figure 1), *L. aucheri* extract have strong activity that reach almost as high as ascorbic acid.

Total antioxidant capacity of *L. sativum* leaves glycosides were evaluated as ascorbic acid equivalents (AAE) per gram of the leaf's glycosides extract. Total antioxidant capacity of the test samples was calculated using the standard line of ascorbic acid (y = 0.0629x + 48.356, $R^2 = 0.4085$). The glycoside compounds content of the test solutions was calculated using the calibration curve of the standard (Figure 1), The results of *L. aucheri* and *L. sativum* was (y = 0.0648x + 48.761, $R^2 = 0.3483$) and (y = 0.0436x + 43.48, $R^2 = 0.3493$) respectively (Figure 2 and 3).

Our results reported that the DPPH radical scavenging activity of the extract from *Lepidium* species increased with increasing concentration (Sandoval *et al.* 2002). This may be due to find some chemical compounds have antioxidant activity, which agreed with (Karimi *et al.*,2011; Vuong *et al.*, 2013).

2- IC 50 assay:

The antioxidant property of various concentration of *L. astivum* and *L. aucheri* was presented by their IC_{50} values, all data was

compared with the IC_{50} value of standard ascorbic acid, and the results presented in (Figure 4).

The greatest DPPH radical scavenging potency of with a minimum IC₅₀ value was recorded for L. aucheri (19.12 µg/mL), followed by ascorbic acid (26.136 μ g/mL) and then L. sativum (149.541µg/mL). The IC₅₀ of L. sativum obtained in our study was higher compared with some studies conducted to assess the antioxidant activity of the methanol extract of L. sativum found that the IC₅₀ values are 62 μ g/ml (Ahmad *et* al., 2015), and Aydemir and Sedabecerik (2011) was IC_{50} value 318.91 ppm for the Turkish L. sativum. The scavenging activity might be due to the different in chemical compounds (Czapecka et al., 2005).Chatoui et al. (2016) recorded 925.22± 0.02 ppm from ethanol extract, while Al-Safi (2019) remember it was 0.146-18.75 mg/ml.

The concentration 1000 μ g/ml showed the highest capacity to neutralize this radical. The lower the IC₅₀ the higher the antioxidant property of a plant. The IC₅₀ values of the different concentrations of leave extracts of *L. sativum* and *L. aucheri*. are presented in the (Figure 4). The *L.aucheri* glycoside compounds exhibited highest antioxidant activity with an IC₅₀ value of 19.12 μ g/mL compared to other fractions. While the value is 26.136 μ g/mL for the standard ascorbic acid (Figure 4).

This free radical scavenging activity might be due to the presence of glycoside compounds in the extracts (Ahamad et al., 2015). The phytochemical compound of Lepidium species can donate hydrogen ions to synthetic free radical compound (DPPH), These may be given rise to the major bioactive compounds. These results supported the observation of some researchers reported that the secondary metabolites provide many species of Lepidium have antimicrobial, anticancer. anti-inflammatory activities, and antioxidant effects (Chatoui et al., 2016; Al-Harbi, 2018; Alqahtani et al., 2019). Some species of Lepidium caused an increase in the antioxidant levels in the blood and organs (Rodriguez-Huaman et al., 2017). Studies have showed that extracts from L. sativum have antibacterial, antioxidant and antiviral effects. these extracts scavenge the reactive oxygen species and radicals and thus protect the cell from oxidative stress (Ait-yahia et al., 2018).

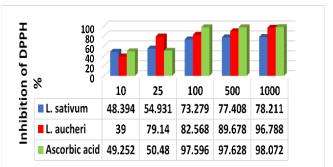


Figure (1) The percentage inhibition of glycosides extracts from *L. sativum* leavesby the antioxidant ascorbic acid.

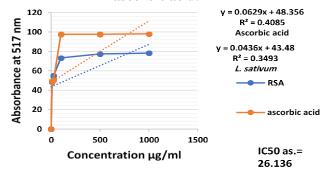


Figure (2) Calibration curve of percentage inhibition of the free radical DPPH by *L.sativum* leaves of glycoside and ascorbic acid.

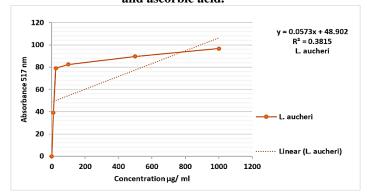


Figure (3) Calibration curve of percentage inhibition of the free radical DPPH by *L.aucheri* leaves of glycoside.

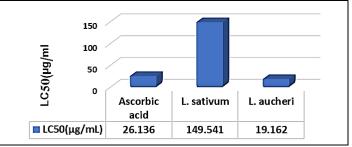


Figure (4): IC50 values of the different glycoside extracts in DPPH scavenging assay.

3- Chemical glycosides composition of *L. sativum* and *L. aucheri* leaves.

The GC-MS chromatogram of *L. satvium* and *L. aucheri* leaves glycosides (Figure 5; Table1).

The GC analysis indicated that L. sativum had high number of glycoside compounds 34 components compared with L. aucheri 21 components. The major components of L. sativum were found to be: Benzyl nitrile (22.24%), N,Nethanol(17.53%),4H-Pyran-4-Dimethylamino one,2,3-dihydro-3, 5 - dihydroxy-6-methyl- (12.48 %); 2-Hydroxy-1-(1'-pyrrolidiyl)-1-buten-3-one (11.08%), D-Proline (7.33%), Butyrolactone (4.97%) and 1-(1'-pyrrolidinyl)-2-propanone (4.14),Benzene,

(isothiocyanatomethyl)(2.05%),Hexadecanoic acid,2-hydroxy-1-(hydroxymethyl)ethylester (1%)

as well as some other compounds were only present in minor amounts. GC -MS analysis of *L. aucheri* represented 21 compounds, DL-Proline, 5-oxo-,methylester(42.26%);2-

Furancarboxaldehyde ,5-methyl-(12.64%);4H-Pyran-4-one, 2,3-dihydro-3,5-dihydroxy-6methyl-(9.87%);2-Piperidineme- thanol (2.86%); Tetrahydropyran 12-tetradecyn-1-ol ether (2.80%); 2-Hydroxy-1-(1'-pyrrolidiyl)-1-buten-3one (2.55%); Hexadecanoic acid, 2-hydroxy-1-(hydroxymethyl) ethyl ester (2.33%); 2-Methoxy-4-vinylphenol (2.02%).

Our results agreed with some research, Malar et al. (2018), which recorded 16.32% of benzyle nitrile, Benzene, (isothiocyanatomethyl) (3.89%), Hexadecanoic acid, 2-hydroxy-1-(hydroxymethyl) ethyl ester (1.14%). As well as Singh and Singh (2017) recorded Hexadecanoic acid, 2-hydroxy-1-(hydroxymethyl) ethyl ester compound (7.13%). Abu-Rumman (2018) recorded 2.93% benzyle nitrile, Benzene, (isothiocyanatomethyl) (0.3%) and Hexadecanoic acid, 2-hydroxy-1-(hydroxymethyl) ethyl ester (10.3%).

study indicated Our that present Butyrolactone compound, this compound identifies in some plants of Mints plants, Butyrolactone has been detected in several foods, such different as pepper (capsicum yellow bell peppers, orange bell annuum), peppers, soybeans, and evergreen blackberries. Butyrolactone is a toxic compound, but it is also used as a pharmacological agent. This study observed that glycerin absents in L. sativum, but it is present in L. aucheri (1.89%). 4H-Pyran-4-one, 2,3-dihydro-3,5-dihydroxy-6-methyland 2-Furancarboxaldehyde, 5-methyl- present in both L. sativum and L. aucheri which agreed with (Al-Safi, 2018). These compounds determined in some species of Brassicaceae (Gopalakrishnan and Udayakumar, 2014). DL-Proline, 5-oxo-, methyl ester, found in some plants (Gopalakrishnan and Udayakumar, 2014).

Natural compounds are also important where they prevent the oxidative stress damage (Boudouda et al., 2015). Hexadecanoic acid, 2hydroxy-1-(hydroxymethyl) ethyl ester is important as antimicrobial, antiallergic, hemolytic, pesticide and antioxidant (Singh and Singh, 2017; Abu-Rumman, 2018). Furan compounds such as (2-Furanmethanol and Furaneol) reported as antioxidant;2-Furancarboxaldehyde, 5-methyl-, 2,4-Dihydroxy-2,5-dimethyl-3(2H)-furan-3one useful as antimicrobial, anti-inflammatory, automatic activity, antiproliferative nerve antioxidant. Glycerin compound (nature of compound sugar alcohol) determined in L. aucheri was used pesticides, herbicidal and antimicrobial; while 4H-Pyran-4-one, 2,3-dihydro-3,5dihydroxy6-methyl- antimicrobial; 2-Methoxy-4vinylphenol, Phenolic compound, antioxidant; L-Proline, 5-oxo-, methyl ester.

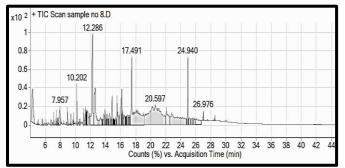


Figure (5) Chromatogram of chemical compounds of *Lepidium sativum leaves*.

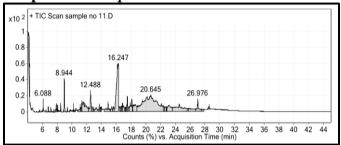


Figure (6) Chromatogram of chemical compounds of *Lepidium aucheri* leaves.

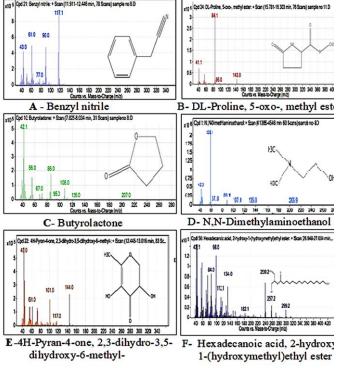


Figure (7) A typical gas chromatogram of the chemical compound of *Lepidium aucheri* leaves.

 Table (1) Chemical constituents of L. satvium and L. aucheri
 leaves glycosides by using GC-MS analysis.

Peak	Chemical constituents	RT	Area	Area %
			%	<i>L</i> .
			<i>L</i> .	sativum
			ucher	
			i	
1	N,N-Dimethylaminoethanol(Deanol; Varesal; Bimanol; DMAE; Kalpur P; Liparon)	4.330	-	17.53
2	Propanoic acid, 2-oxo-, methyl ester	5.060	-	0.76
3	Pyrazine, methyl-	5.623	-	0.88
4	2-Furanmethanol	6.762	1.54	1.36
5	4-Cyclopentene-1,3-dione	7.117	1.90	1.51
6	N-Ethylidene t-butylamine	7.478	-	1.23
7	Pyrazine, 2,6-dimethyl-	7.804	-	1.13
8	Butyrolactone	7.957	1.63	4.97
9	2-Furancarboxaldehyde, 5-methyl-	8.937	12.64	1.55
10	2,4-Dihydroxy-2,5-dimethyl-3(2H)- furan-3-one	9.396	1.62	0.94
11	Pyrazine, trimethyl-	9.771	-	1.06
12	1-(1'-pyrrolidinyl)-2-propanone	10.202	1.41	4.14
13	4H-1,2,4-Triazole, 4-methyl-	11.105	-	1.78
14	2,5-Piperazinedione, 3-methyl-6-(1- methylethyl)-	11.348	-	1.96
15	4H-Pyran-4-one, 2,3-dihydro-3,5- dihydroxy-6-methyl-	11.494	9.87	12.48
16	Benzyl nitrile (Benzyl cyanide; Phenylacetonitrile)	12.286	1.33	22.24
17	Pyrazine, 2-ethyl-5-methyl-	14.524	-	1.65
18	2-Methoxy-4-vinylphenol	14.871	2.02	3.05
19	Benzene, (isothiocyanatomethyl)-	15.517	-	2.05
20	Tridecanenitrile	15.594	-	1.11
21	3-Pyridinamine, N,N-dimethyl-	15.733	-	0.69
22	Pyrrolidine, 1-(1-cyclohexen-1-yl)	16.129	-	1.74
23	Benzeneacetamide	16.282	-	0.79
24	2-Hydroxy-1-(1'-pyrrolidiyl)-1-buten- 3-one	17.491	2.55	11.08

25	5,10-Diethoxy-2,3,7,8-tetrahydro-	22.070	-	1.31
	1H,6H-dipyrrolo[1,2-a:1',2'-d]pyrazine			
26	D-Proline	24.948	-	7.33
27	Hexadecanoic acid, 2-hydroxy-1-	26.976	2.33	1.00
	(hydroxymethyl)ethyl ester			
28	4- (2,5-Dihydro-3-	14.524	-	1.58
	methoxyphenyl)butylamine			
29	Cyclohexanone, 2-(2-butynyl)-	14.871	-	2.05
30	Benzene, (isothiocyanatomethyl)-	15.517	-	2.05
31	Pyrrolidine, 1-acetyl-	12.766	-	0.69
	Azocine, octahydro-			
32	Pyrrolidine, 1-(1-cyclohexen-1-yl)-	16.129	-	1.74
33	Benzeneacetamide	16.282	-	0.79
35	5,10-Diethoxy-2,3,7,8-tetrahydro-	22.070	-	1.31
	1H,6H-dipyrrolo[1,2-a:1',2'-d]pyrazine			
36	2-Piperidinemethanol	6.088	2.98	-
37	Acetic acid, 2-(dimethylamino) ethyl	8.458	1.50	-
	ester			
38	Cyclopentane, 1-methyl-2-(2-	11.091	1.43	-
	propenyl)-, trans-			
39	Furaneol	11.313	2.19	-
40	Glycerin	11.703	1.89	-
41	Tetrahydropyran 12-tetradecyn-1-ol	13.683	2.80	-
	ether			
42	DL-Proline, 5-oxo-, methyl ester	16.122	42.26	-
43	Pyrrolidin-2-one, 5-pentyl-	16.796	1.41	-
44	4-Methyl-2,5-	18.054	1.84	-
	dimethoxybenzaldehyde			
45	1-Amino-4,6-dimethyl-2-oxo-	24.460	2.86	-
	1,2-dihydropyridine-3-			
	carbonitrile			
			100	100
		1	100	100

IV.

V. CONCLUSION

present study, the identified In the phytocompounds of two species of Lepidium with molecular formula and structure, Thus the GC-MS analysis is the first step towards understanding the nature of active principles in Lepidium species. This study has been good source to produce many modern drugs and drug development (Ravi et al., 2018). The two species of Lepidium have presence biological chemicals of active such as isothiocyanates phytochemical and some components such as benzyle nitrile, furan and glycerin compound, the presence of phytochemicals leaves of L. sativum may be responsible for controlling diseases.

Natural products and biological effects of *Lepidium* species are reported in this study, the phytochemical study of the Leaves of this species has led to the separation and identification and we found the DPPH is good exhibited. The results of this study indicate that glycoside extract of *L. sativum* leaves possess significant antioxidant properties,

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