

Preparation and Characterization of New Heterocyclic Derivatives, Six and Five Ring from Acetanilide

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Abstract-The aim of the study synthesis of new heterocyclic derivatives (six and five ring) from acetanilide. Firstly the chalcones have been synthesized from the reaction acetanilide with aromatic aldehyde [2-nitrobenzaldehyde, 4- dimethylaminobenzaldehyde, 2-naphthaldehyde] in dilute solution of ethanolic sodium hydroxide at room temperature. This reaction occurs depending on Claisen Schmidt condensation, then a new oxazine and thiazine derivatives (six ring) have been synthesized by the reaction chalcones with urea and thiourea. Finally the synthesized chalcones were reacted with hydrazine hydrate to produce new pyrazol derivatives (five ring). All derivatives were characterized by physical properties, FTIR, ¹H-NMR and ¹³C-NMR spectra.

Keywords- acetanilide, chalcones, thiazine, oxazine, pyrazol.

I. Introduction

Acetanilide is a solid, no having odor chemical and have a peel in appearance. It is also named N-phenylacetamide, acetanilid, acetanil and in the past known the commerce name antifebrin. Acetanilide is applied to a curb of hydrogen peroxide decompositions. It also uses key intermediate for the industry of the sulfa drugs, rubber synthesis, dyes and camphor structure (Baty *et al.*, 1988). Chalcones are appear the important in biosynthesis and can preparation different compounds as pyrazoline, thiazine (Farooque and Zulfequar, 2012), pyrimidine to have heterocyclic ring (Ishwar, 2014). They have α , β -unsaturated keton to responsible for antioxidant (Kamble, 2011), anti-inflammatory (Hsieh, 2000) and antibacterial activity (Davinder, 2012). Oxazine is heterocyclic compound (six ring) contain N and O in structure, can be prepare form chalcone and posses a large activity such as antimicrobial (Chaitra and Rohini, 2018), anticancer (Ouberai *et al.*, 2006), antiviral Anushal *et al.*, (2015), antimalarial (Beena, 2013), anticoagulant activity (Sawant *et al.*, 2012) and antifungal (Sayaji and Pravina, 2013). Thiazine is heterocyclic compound (six ring) contain N and S in structure. It exhibit wide interest to bioactive like antibacterial (Naeemah *et al.*, 2018), hypoglycemic (Alok *et al.*, 2015) and anti-inflammatory (Nagaraj and Reddy, 2008). Pyrazol heterocyclic derivatives (five ring) have two nitrogen in membered ring (Rana *et al.*, 2018). It and its derivatives

display important of biological activity like anti-depressant, anti-convulsant (Ozdemir *et al.*, 2007), antiamebic (Budakoti *et al.*, 2006), antibacterial (Korablina *et al.*, 2016).

II. Experimental Methods

1-Synthesis of Chalcones [A₁-A₃] (Choudhary *et al.*, 2011)

Acetanilide (0.01 mole) of dissolved in ethanol (10 ml), then added aqueous sodium hydroxide 40 % (10 ml) and mixed for (25-30 min). After that aromatic aldehyde [m-nitrobenzaldehyde, para-dimethylaminobenzaldehyde, 2-naphthaldehyde] (0.01 mole) was added and mixed for 20 hours at room temperature. After complete this reaction added acetic acid to neutralized solution then put it in crashed ice and precipitated it then filtered the crystal material and recrystallization from ethanol. Table [1] shows the physical data and FT.IR of compounds [A₁-A₃].

2- Synthesis of Oxazine and Thiazine Derivatives [B₁-B₃, C₁-C₃] (Zainab, 2014)

Chalcone [A₁-A₃] (0.01mol) and urea, thiourea (0.01mol) were dissolve it in ethanolic sodium hydroxide (25 ml) was stirred with a magnetic stirrer and refluxed 2-3 hours. Cold distilled water was added with continuous stirring for 1 hour then kept in refrigerator for 20 hours. The precipitated compound was filter and recrystallization from ethanol. Table [2] shows the physical data and FT.IR of compounds [B₁-B₃, C₁-C₃].

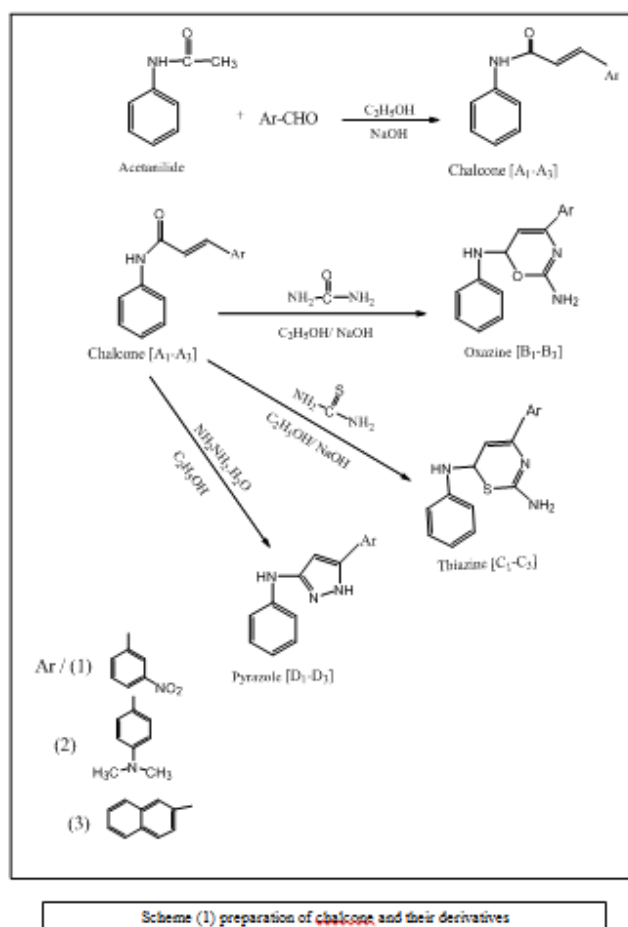
2- Synthesis of Pyrazole Derivatives [D₁-D₃] (Zainab, 2014)

Taken from of chalcone [A₁-A₃] (0.01 mol) and hydrazine hydrate (0.01 mol) in ethanol (20 ml) was refluxed for 5 hours. Add of cold distilled water and stirring 1 hour. The precipitation was filtered, then take the crystal and recrystallization from ethyl alcohol. Table [3] shows the physical data and FT.IR of compounds [D₁-D₃].

III. Results and Discussion

New heterocyclic derivatives (six and five ring) from acetanilide have been synthesized. In the begain step chalcones [A₁-A₃] was prepared from the reaction acetanilide with aromatic aldehyde in dilute solution of ethanolic sodium hydroxide in room temperature this reaction occurs depending on Claisen Schmidt condensation, then you synthesized new oxazine and thiazine derivatives [B₁-B₃, C₁-C₃] by the reaction chalcones

with urea and thiourea. Finally chalcones were reacted with hydrazine hydrate to produce new pyrazol derivatives [D₁-D₃], scheme [1]. FTIR spectra of derivatives [A₁-A₃] appear absorption bands at [(3022-3054) cm⁻¹ to (C-H) aromatic, (3334-3456) cm⁻¹ to (N-H) and (1639-1662) cm⁻¹ to (C=O)] (Silverstein *et al.*, 2005), other data display in table [1].



FTIR spectra of derivatives [B₁-B₃, C₁-C₃] show disappearance of absorption band at (1639-1662) cm⁻¹ which belong to the carbonyl group and appear of absorption band at (1610-1632) cm⁻¹ due to (C=N) and bands [(3214-3304) sym, (3371-3483) asym] to (NH₂), other data were display in table [2]. ¹H-NMR of oxazine [B₂] appear signals at [δ(6.1-7.4) ppm (m, Ar-H), δ 8.7 ppm (s, 2H, NH₂), δ 4.5 ppm (s, H, NH) and δ 3.2 ppm (s, 3H, 2CH₃)]. ¹³C-NMR of compound [B₂] notice peaks aromatic carbon at δ(113-135) ppm, δ(150) ppm due to (C-N), δ(60) ppm to (C-O) and δ(45) ppm to (CH₃) group. ¹H-NMR of thiazine [C₁] display signals at [δ(6.43-7.81) ppm (m, Ar-H), δ 8.3 ppm (s, 2H, NH₂) and δ 4.2 ppm (s, H, NH)] (Rahman and Choudhary, 1991) ¹³C-NMR of compound [C₁] notice peaks aromatic carbon at δ(122-139) ppm, δ(148) ppm due to (C-N) and δ(35) ppm to (C-S).

FTIR spectra of derivatives [D₁-D₃] appearance of absorption band at (1615-1639) cm⁻¹ belong to (C=N) and bands (3300-3354) cm⁻¹ due to (N-H), the other data were display in table [3]. ¹H-NMR of pyrazol [D₃] display signals at [δ(6.89-8.11) ppm (m, Ar-H), δ 11.9 ppm (s, H, NH pyrazol) and δ 3.9 ppm (s, H, NH)]. ¹³C-NMR of compound [D₃] appear peaks of aromatic carbon at δ(117-128) ppm and δ(156) ppm due to (C-N).

Table (1). Physical data and FTIR data of compounds [A₁-A₃]

Comp. No.	Compound Structure	M.P. °C	Yield %	Color	FTIR Absorptions cm ⁻¹				
					ν (C-H) aromatic	ν(N-H)	ν (C=O)	ν (C=C) Aromatic	Others
A ₁		62-64	72	Yellow	3031	3334	1662	1552	ν(NO ₂) = 1371
A ₂		54-56	90	Brown	3022	3342	1641	1558	ν(C-H) aliphatic = 2974
A ₃		93-97	82	Off white	3054	3356	1639	1550	-

Table (2). Physical data and FTIR data of compounds [B₁-B₃] and [C₁-C₃]

Comp. No.	Compound Structure	M.P. °C	Yield %	Color	FTIR Absorptions cm ⁻¹				
					ν (C-H) aromatic	ν(N-H) aq. sym.	ν (C=N)	ν (C=C) aromatic	Others
B ₁		195-197	70	Off white	3014	3429 3266	1620	1590	ν(NO ₂) = 1266 ν(C-O) = 1126
B ₂		172-175	84	white	3034	3371 3216	1622	1581	ν(C-H) aliphatic = 2891 ν(C-O) = 1199
B ₃		278-280	80	Off white	3016	3420 3252	1612	1571	ν(C-O) = 1156
C ₁		118-120	83	Brown	3060	3482 3292	1628	1570	ν(NO ₂) = 1212 ν(C-O) = 1229
C ₂		181-182	58	white	3055	3485 3275	1618	1581	ν(C-H) aliphatic = 2909 ν(C-O) = 1256
C ₃		242-245	56	white	3011	3456 3290	1610	1592	ν(C-O) = 1279

Table (3). Physical data and FTIR data of compounds [D₁-D₃]

Comp. No.	Compound Structure	M.P. °C	Yield %	Color	FTIR Absorptions cm ⁻¹				
					ν (C-H) aromatic	ν(N-H)	ν (C=N)	ν (C=C) aromatic	Others
D ₁		240-262	71	Yellow	3061	3354	1623	1590	ν(NO ₂) = 1301 ν(C-S) = 1265
D ₂		190-192	63	Brown	3049	3331	1639	1594	ν(C-H) aliphatic = 2910 ν(C-S) = 1270
D ₃		210-212	59	Brown	3070	3300	1615	1592	ν(C-S) = 1264

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